

Science Long Term Map Chemistry y11

Subject Intent/Aims

Expose all students to a broad range of learning opportunities to deepen their knowledge and understanding of themselves and the world around them and to build a solid foundation of Science knowledge and skills. We believe in developing curiosity and understand that science is an active process with many questions to be answered and still to be asked. We provide an understanding of how knowledge was derived, discovered and came to be accepted by the scientific community. By focusing on thinking, interpreting and evaluating rather than simply memorising scientific fact we intend to enable our students to use the skills that they need to answer their own scientific questions.

Our focus on the scientific process as a way of thinking and working will allow our students to develop their own ideas, attitudes and interpretations.

Topic Rate and extent of reactions National Curriculum: Rate and extent of chemical change • factors that influence the rate of reaction: varying temperature or concentration, changing the surface area of a solid reactant or by adding a catalyst • factors affecting reversible reactions. GCSE Chemistry Specification Specification for first teaching in 2016 (aqa.org.uk)	Topic Organic National Curriculum: carbon compounds, both as fuels and feedstock, and the competing demands for limited resources • fractional distillation of crude oil and cracking to make more useful materials • bonding of carbon leading to the vast array of natural and synthetic organic compounds that occur due to the ability of carbon to form families of similar compounds, chains and rings GCSE Chemistry Specification Specification for first teaching in 2016 (aga.org.uk)	formation • evidence, and uncertainties in	assess environmental impacts associated	Chemical analysis National Curriculum: Chemical analysis • distinguishing between pure and impure substances • identification of common gases Identification of common anions and cations GCSE Chemistry Specification Specification for first teaching in 2016 (aga.org.uk)	
Composition	Composition	Composition	Composition	Composition	
Rates of reaction	Hydrocarbons	· ·	Recognise the properties and uses of	Understand how scientists use	
Factors affecting rate of reaction	Fractional distillation Uses of and cracking of crude oil	and the reasons for and consequences of the changes	Leramics, composites and polymers. Explain the issues around corrosion	Chemical analysis to detect specific chemicals	
Measuring rate of reaction	Alkenes Reactions of alkenes		Explain the differences between Finite and renewable resources		
Rates experiments			Describe Potable water and link to water		
Rate of reaction graphs			treatment		
Reversible reactions					
Le Chatelier's principle					
Components	Components	Components	Components	Components	







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Component 1: Know ways of speeding up reactions. Component 2: Know Le Chatelier's principle Component 3: Know the variables and hazards in an investigation. Component 4: Know how to plan and carry out investigations to test different hypotheses.	Component 1: Know the process of fractional distillation and cracking of crude oil Component 2: Know the difference between individual alkanes and alkenes. Component 3: Know how organic molecules can be modified to produce new and useful materials such as polymers.	atmosphere. Component 2: know why data needed to answer scientific questions may be uncertain, incomplete or unavailable. Component 3: know how the atmosphere has changed over geological timescales. Component 4:know the environmental implications of greenhouse gas emissions	produce potable water and to treat waste water. Component 3: know the impact of products	Component 1: know what is meant by 'pure substances' and how they can be distinguished from mixtures. Component 2: know how instrumental methods are used to identify substances. Component 3: know the different tests used to test gases such as hydrogen, oxygen carbon dioxide and chlorine
composites	composites	composites	composites	composites
predict the effects of changing temperature, pressure and concentration on equilibrium systems whilst exploring reversible reactions use collision theory and ideas about activation energy to make predictions	Explain the use of fractional distillation as a separation technique Relate the properties of the fractions to their structure Explain the financial benefits of cracking	Suggest how the atmosphere has changed due to chemical processes Analyse data to formulate conclusions about atmospheric changes Debate issues around the greenhouse effect Suggest solutions to reduction of atmospheric pollution	Investigate water samples to analyse salt content and potability Complete a life cycle assessment on a range of products Evaluate the uses of different materials	Compare the use of instrumental methods with wet analysis methods Identify gases using recognised gas tests
recognise and describe patterns and trends in data use models and data to make predictions and communicate findings and reasoned conclusions				
Higher Order Knowledge	Higher Order Knowledge	Higher Order Knowledge	Higher Order Knowledge	Higher Order Knowledge
Know how to calculate the gradient of a tangent to the curve on these graphs as a measure of rate of reaction at a specific time. Know the relative amounts of all the reactants and products at equilibrium depend on the conditions of the reaction. If a system is at equilibrium and a change is made to any of the conditions, then the system responds to counteract the change. The effects of changing	Know the functional groups of organic compounds: Addition polymers Alcohols Carboxylic acids Condensation polymers Naturally occurring polymers Know natural polymers	Evaluate the quality of evidence in reports about global climate change and learn how peer review works.	know alternative methods to extract a metal from low-grade ores that avoid the environmental impact of mining. Use LCAs to evaluate the uses of different materials for products Know how the Haber process is useful as a feedstock for fertilizers	Know the chemical tests for anions and cations and be able to use them qualitatively to identify unknown compounds







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Department	Planning	2025	/2026
Department	1 10111111115	2023	, 2020

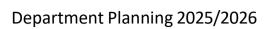
conditions on a system at equilibrium can be predicted using Le Chatelier's Principle.				
Know the effects of changing conditions in the Haber Process				
<u>Key terms</u>	<u>Key terms</u>	Key terms	<u>Key terms</u>	<u>Key terms</u>
Activation Energy	Finite Resource	Atmosphere	Finite Resource	Pure Substance
Enzymes	Biomass	Carbon Footprint	Renewable Resource	Formulation
Closed System	Hydrocarbon	Fossil Fuels	Sustainable Development	Cation
Dynamic Equilibrium	Displayed Formula	Sulfur Impurities	Life Cycle Assessment	Anion
Le Chatelier's Principle	Homologous Series	Greenhouse Gases	Potable Water	Precipitate
Turbidity	Alkanes	Global Dimming	Desalination	Compound
Catalyst	Saturated	Incomplete Combustion	Ore	Element
	Fractional Distillation	Nitrogen Oxides	Alloy	Impure
	Fraction	Particulate	Corrosion	Melting Point
	Complete Combustion		Rusting	Mixture
	Incomplete Combustion		Sacrificial Protection	Chromatography
	Flammability		Bioleaching	Stationary Phase
	Viscosity		Leachate	Mobile Phase
	Cracking		Leaching	Chromatogram
	Alkenes		Phytomining	Rf Value
	Unsaturated		Thermosetting Polymer	
	Functional Group		Thermosoftening Polymer	
	Addition Reaction			
	Hydrogenation			
	Hydration			
	Alcohols			
	Carboxylic Acids			
	Esters			
Final Composition/Deliberate Practice	Final Composition/Deliberate Practice	Final Composition/Deliberate Practice	Final Composition/Deliberate Practice	Final Composition/Deliberate Practice
Planning, carrying out and analysing an investigation	Complete a series of 6 mark questions related to the industrial processes	Debate the benefits of alternative fuels on global warming	Required practical investigating the purity of water samples	Planning, carrying out and analysing an investigation
Required practical: measuring rate of	. c. s. c.			RP chromatography
reaction using different methods				TRIPLE analysis investigation
Summative/Formative assessment	Summative/Formative assessment	Summative/Formative assessment	Summative/Formative assessment	
Core questions RRR to recall prior knowledge will be tested at the beginning of each lesson and self- assessed by the student. Century nuggets and PPQs LC for collision theory	will be tested at the beginning of each lesson and self- assessed by the student. Century nuggets and PPQs LC for organic chemistry	will be tested at the beginning of each lesson and self- assessed by the student. Century nuggets and PPQs LC for evolution of ther atmosphere	will be tested at the beginning of each lessor and self- assessed by the student. Century nuggets and PPQs LC for water purification	will be tested at the beginning of each lesson and self- assessed by the student. Century nuggets and PPQs LC for chemical analysis
End of unit assessment will be marked with	End of unit assessment will be marked with	End of unit assessment will be marked with	End of unit assessment will be marked with	End of unit assessment will be marked with







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Numeracy	Literacy	Numeracy	Literacy	Numeracy	Literacy	Numeracy	Literacy	Numeracy	Literacy	
Maths skills – graphs, calculations	English – literacy skills – focusing on keywords, tier 3 vocabulary, connectives, SPAG, synonyms,	Maths skills – graphs, calculations	English – literacy skills – focusing on keywords, tier 3 vocabulary, connectives, SPAG, synonyms,	Maths skills – graphs, calculations	English – literacy skills – focusing on keywords, tier 3 vocabulary, connectives, SPAG, synonyms,	Maths skills – graphs, calculations	English — literacy skills — focusing on keywords, tier 3 vocabulary, connectives, SPAG, synonyms,	Maths skills – graphs, calculations	English – literacy skills – focusing on keywords, tier 3 vocabulary, connectives, SPAG, synonyms,	
Cross curricular links		Cross curricular links		Cross curricular links		Cross curricular links				
History- Fritz Haber a concentration camps Biology- Use of enzy Food technology- Us food production and	s. mes in reactions. e of enzymes in	Biology- Structure an molecules. Food Technology- Us production. DT- Use of materials, new materials. Geography- Use of pl problems this creates	e of esters in food development of astics and the	Geography- Climate human activity on th		Engineering – reuse			Food technology – analysis of additives in foods	
SMSC British Value		tish Value			RSHE					
learning and a range of The high expectations pupils will regularly be Pupils are expected to and appreciate others The majority of topics	There will be multiple opportunities for students develop spiritually; being creative in their searning and a range of activities The high expectations placed on the student from the school and department mean that pupils will regularly be made aware of the right and wrong morally or the different topics but also show respect and appreciate others in the classroom. The majority of topics will give the students opportunity to develop their social skills, from siving presentations to working in group tasks. Students will be taught the legal implications of using the interest to Students will be taught to fully appreciate other students view importance of being respectful Students will be taught the importance of selecting valid infor sources for any presentation tasks that they do. Students are taught how to contribute to life in modern Britain history of scientific discovery Students will be taught the legal implications of using the interest to students will be taught to fully appreciate other students view importance of being respectful Students will be taught the legal implications of using the interest to students will be taught to fully appreciate other students view importance of being respectful Students will be taught the legal implications of using the interest to fully appreciate other students view importance of being respectful Students will be taught the legal implications of using the interest to fully appreciate other students view importance of being respectful Students will be taught the legal implications of using the interest to fully appreciate other students view importance of being respectful Students will be taught the legal implications of using the interest to fully appreciate other students view importance of being respectful Students will be taught to fully appreciate other students view importance of selecting valid information to fully appreciate other students view importance of selecting students will be taught to fully appreciate other students view importance of select		ciate other students viewpoin te of selecting valid informati at they do. e to life in modern Britain by	ion from reliable learning about the	may arise.	ware of online relationship	os and the sexual issues that ical and mental health when			





Adapted Curriculum Content: Adapted Curriculum Content:		Adapted Curriculum content:	Adapted curriculum content:	Adapted curriculum content:	
	c ombined: limited to alkanes, fractional distillation and cracking.	Thers are no adaptations in this unit	Phytomining and bioleaching triple only	Anion and cation analysis triple only	
Adaptive Implementation Practices: Coloured paper/pens	Adaptive Implementation Practices: Coloured paper/pens	Adaptive Implementation Practices: Coloured paper/pens	Adaptive Implementation Practices: Coloured paper/pens	Adaptive Implementation Practices: Coloured paper/pens	
Differentiated worksheets	Differentiated worksheets	Differentiated worksheets	Differentiated worksheets	Differentiated worksheets	
Differentiated worksheets	Differentiated tasks	Differentiated tasks	Differentiated tasks	Differentiated tasks	
Seating plans to maximise	Seating plans to maximise	Seating plans to maximise	Seating plans to maximise	Seating plans to maximise	
concentration allowing for	concentration allowing for	concentration allowing for	concentration allowing for	concentration allowing for	
visual/hearing impairments etc	visual/hearing impairments etc	visual/hearing impairments etc	visual/hearing impairments etc	visual/hearing impairments etc	
Appropriate use of IWB	Appropriate use of IWB	Appropriate use of IWB	Appropriate use of IWB	Appropriate use of IWB	
Dual coding	Dual coding	Dual coding	Dual coding	Dual coding	
Spare equipment	Spare equipment	Spare equipment	Spare equipment	Spare equipment	
Modelling experimental detail	Modelling experimental detail	Modelling experimental detail	Modelling experimental detail	Modelling experimental detail	
Pre drawn tables/graphs/diagrams to	Pre drawn tables/graphs/diagrams to	Pre drawn tables/graphs/diagrams to	Pre drawn tables/graphs/diagrams to	Pre drawn tables/graphs/diagrams to	
be labelled	be labelled	be labelled	be labelled	be labelled	















